MATERIAL SAFETY DATA SHEET

National Institute of Standards and Technology **Standard Reference Materials Program** 100 Bureau Drive, Stop 2320

Gaithersburg, Maryland 20899-2320

SRM Number: 3003 MSDS Number: 3003

SRM Name: o-Xylene in Methanol

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SECTION I. MATERIAL IDENTIFICATION

Material Name: o-Xylene in Methanol

Description: SRM 3003 consists of two 5-milliliter sealed borosilicate glass ampoules containing approximately 2.5 mL of a solution of o-xylene in methanol.

Other Designations: o-Xylene (1,2-dimethylbenzene; o-dimethylbenzene; o-methyltoluene; ortho-xylene; 1,2-xylene; xylene, o-xylol) in Methanol (methyl alcohol; wood alcohol; methyl hydroxide; carbinol; monohydroxymethane; wood spirit; wood naphtha; methylol)

Chemical Formula CAS Registry Number Name CH₃OH Methanol 67-56-1 95-47-6 o-Xylene C_8H_{10}

DOT Classification: Methanol; UN1230; Packing Group II; Hazard Class 3.

SECTION II. HAZARDOUS INGREDIENTS

Hazardous Components	Nominal Concentration (%)	Exposure Limits and Toxicity Data	
Methanol	99	OSHA TWA: 260 mg/m³ (200 ppm)	
		NIOSH recommended TWA (skin): 260 mg/m³ (200 ppm) (10 h)	
		NIOSH recommended STEL (skin): 325 mg/m³ (250 ppm)	
		OES, UK TWA (skin): 266 mg/m ³ (200 ppm)	
		OES, UK STEL (skin): 333 mg/m ³ (250 ppm)	
		Human, Inhalation TC _{LO} : 86 000 mg/m ³	
		Human, Oral LD _{LO} : 143 mg/kg	
		Man, Oral TD _{LO} : 3 429 mg/kg	
o-Xylene	1	ACGIH TWA: 100 ppm	
		ACGIH STEL: 150 ppm	
		OSHA TWA: 435 mg/m ³ (100 ppm)	
		NIOSH recommended TWA: 435 mg/m³ (100 ppm) (10 h)	
		NIOSH recommended STEL: 655 mg/m³ (150 ppm)	
		UK OES TWA (skin) (mixed isomers): 220 mg/m ³ (50 ppm)	
		UK OES STEL (skin) (mixed isomers): 441 mg/m³ (100 ppm)	
		Rat, Oral LD ₅₀ : 3 617 mg/kg	
		Human, Inhalation LC _{LO} : 6 125 ppm/12 h	

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SECTION III. PHYSICAL/CHEMICAL CHARACTERISTICS

Methanol	o-Xylene		
Appearance and Odor: a clear, colorless liquid with a characteristic alcoholic odor	Appearance and Odor: a clear, colorless liquid with a sweet odor		
Relative Molecular Mass: 32.04	Relative Molecular Mass: 106.17		
Density: 0.7914 g/m ³	Density: 0.8802 g/m ³		
Boiling Point: 65 °C (149 °F)	Boiling Point: 144 °C (291 °F)		
Freezing Point: -94 °C (-137 °F)	Freezing Point: -25 °C (-13 °F)		
Vapor Pressure (@ 20 °C): 97.25 mmHg	Vapor Pressure (@ 25 °C): 5.2 mmHg		
Evaporation Rate (butyl acetate = 1): 4.6	Evaporation Rate (butyl acetate = 1): 0.7		
Viscosity (@ 20 °C): 0.59 cP	Viscosity (@ 20 °C): not available		
Solubility in Water: soluble	Solubility in Water (@ 20 °C): 0.0175 %		
Solvent Solubility: soluble in ether, benzene, alcohol, acetone, chloroform, ethanol, ketones, and most organic solvents	Solvent Solubility: soluble in alcohol, ether, benzene, acetone, and organic solvents		

NOTE: The physical and chemical data provided are for the pure components. Physical and chemical data for this o-xylene/methanol solution do not exist. The actual behavior of the solution may differ from the individual components.

SECTION IV. FIRE AND EXPLOSION HAZARD DATA

Methanol

Flash Point: 11 °C Method Used: Closed Cup Autoignition Temperature: 385 °C

Flammability Limits in Air (Volume %): UPPER: 36

LOWER: 6.0

o-Xylene

Flash Point: 17 °C Method Used: Not available. Autoignition Temperature: 463 °C

Flammability Limits in Air (Volume %): UPPER: 6.7

LOWER: 0.9

Unusual Fire and Explosion Hazards: Methanol and o-Xylene are a severe fire and explosion hazards when exposed to heat or flame. Vapors are heavier than air and may travel a considerable distance to a source of ignition and flash back. Vapor and air mixtures are explosive.

Extinguishing Media: Use alcohol-resistant foam, dry chemical, carbon dioxide, or water spray.

Special Fire Procedures: Fire fighters should wear a self-contained breathing apparatus (SCBA) with a full face piece in the pressure demand or positive mode and other protective clothing.

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SECTION V. REACTIVITY DAT	A			
Stability:	X Stable	Unstable		
Stable at normal temperature				
		t, sparks, flames, or other source act with the skin. Keep out of v		
		material is incompatible with h	alo carbons, con	nbustible materials,
See Section IV: "Unusual F	ire and Explosion I	Hazards".		
Hazardous Decomposition and various organic organic		Thermal decomposition products	may include tox	ic oxides of carbon
Hazardous Polymerization	: Will (Occur X Will Not Oc	cur	
SECTION VI. HEALTH HAZARI	D DATA			
Route of Entry: X	X Inhalation	X Skin	X_ Ingestion	
absorbed through skin. Ing sensation, coughing, wheez cause damage to the eyes, I convulsions. o-Xylene: o-Xylene may be irritating to the eyes, mucou occur by inhalation of vapirritation and central nervo euphoria, headache, nauses prolonged inhalation of vapo occur if aspirated into the lu	estion may be fatal ing, laryngitis, sho liver, heart, and kid the harmful by inhalates membranes, and upors at 200 ppm. The system excitation at the control of the contro		on. As a vapor on of the upper reconcentrations d respiratory differences, and anore acute exposure.	ay include burning ng. Exposure can al disturbances and or mist, o-xylene is espiratory tract may may cause severe ficulty, chest pain, xia. Repeater or Lung damage may
		by Exposure: o-Xylene may orders, skin disorders, and allers		al nervous system.
Target Organ(s) of Atta	ck: Central nervou	us system (CNS).		
Listed as a Carcinogen/Pot	tential Carcinogen	(Methanol):	Yes	No
In the National Toxicol	ogv Program (NTP)) Report on Carcinogens	Tes	X
In the International Age		X		
By the Occupational Sa		X		
Listed as a Carcinogen/Pot	tential Carcinogen	(o-Xylene):	V 7	NI.
In the National Toyical	ogy Program (NTD)) Report on Carcinogens	Yes	No X
		n Cancer (IARC) Monographs		$\frac{X}{X}$
By the Occupational Sa				$\frac{X}{X}$

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EMERGENCY AND FIRST AID PROCEDURES:

Skin Contact: Remove contaminated shoes and clothing. Rinse affected area with large amounts of water followed by washing the area with soap and water. Watch for chemical irritations and treat them accordingly. Obtain medical assistance if necessary.

Eye Contact: Immediately flush eyes, including under the eyelids, with copious amounts of water for at least 15 minutes. Obtain medical assistance.

Inhalation: If inhaled, move the victim to fresh air. If breathing is difficult, give oxygen; if the victim is not breathing, give artificial respiration by qualified personnel. Obtain medical assistance if necessary.

Ingestion: If ingested, wash out mouth with water. Obtain medical assistance immediately.

SECTION VII. PRECAUTIONS FOR SAFE HANDLING AND USE

Steps to be Taken in Case Material Is Released or Spilled: Notify safety personnel of major spills and/or leaks. Evacuate nonessential personnel. Avoid heat, flames, sparks, and other sources of ignition. Stop the leak if one can do so without risk. Absorb small spills with sand or other non-combustible absorbent material and place into containers for proper disposal.

Waste Disposal: Follow all federal, state, and local laws governing disposal. Methanol is subject to disposal regulations U.S. EPA 40 CFR 262, Hazardous Waste Number U154. Keep o-Xylene out of sewers and water supplies. o-Xylene is subject to disposal regulations U.S. EPA 40 CFR 262, Hazardous Waste Number U239.

Handling and Storage: Persons handling this material must wear protective eyewear, clothing, and gloves to prevent contact with this material. Methanol and o-xylene are subject to storage regulations U.S. OSHA 29 CFR 1910.106. Keep separated from incompatible substances.

NOTE: Contact lenses pose a special problem; soft lenses may absorb irritants and all lenses concentrate them. **DO NOT** wear contact lenses in the laboratory.

This material should be stored in the dark at temperatures between 10 °C and lower than 30 °C, in a well-ventilated area away from incompatible materials and conditions. Protect containers from physical damage.

SECTION VIII. SOURCE DATA/OTHER COMMENTS

Sources: MDL Information Systems, Inc., MSDS *o-Xylene*, 17 June 2004.

MDL Information Systems, Inc., MSDS Methyl Alcohol, 16 September 2004.

Disclaimer: Physical and chemical data contained in this MSDS are provided only for use in assessing the hazardous nature of the material. The MSDS was prepared carefully, using current references; however, NIST does not certify the data on the MSDS. The certified value for this material is given in the NIST Certificate of Analysis.

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